

## Publications List – Haim Gizbar

Peer reviewed journals

1. Development of novel rechargeable non-aqueous magnesium batteries. D. Aurbach, L. Zhongua, A. Schechter, Y. Gofer, **H. Gizbar**, R. Turgeman, Y. Cohen, M. Moshkovich and E. Levi, *Nature*, 407, 724 (2000).
2. Electrolyte solutions for rechargeable magnesium batteries based on organo magnesium chloro aluminate complexes. D. Aurbach, **H. Gizbar**, A. Schechter, O. Chusid, H.E. Gottlieb, Y. Gofer, and I. Goldberg, *J. Electrochem. Soc.*, 149, A115 (2002).
3. A comparison between the electrochemical behavior of reversible magnesium and lithium electrodes. D. Aurbach, Y. Gofer, A. Schechter, O. Chusid, **H. Gizbar**, Y. Cohen, M. Moshkovich, and R. Turgeman, *J. Power Sources* 97-98, 269 (2001).
4. A short review on the comparison between Li battery systems and rechargeable Mg battery technology. D. Aurbach, L. Zhongua, a. Schechter, Y. Gofer, **H. Gizbar**, R. Turgeman, Y. Cohen, M. Moshkovich, and E. Levi, *J. Power Sources* 97-98, 28 (2001).
5. Solid state rechargeable magnesium batteries, O. Chusid, Y. Gofer, **H. Gizbar**, Y. Vestfrid, E. Levi, and D. Aurbach, *Advanced Materials* 15, 627 (2003).
6. Comparative kinetic and thermodynamic studies of  $Mg^{2+}$  and  $Li^+$  ion insertions into the  $Mo_6S_8$  Chevrel phase as a function of solvent nature and temperature, M.D. Levi, E. Lancry, **H. Gizbar**, Z. Lu, E. Levi, Y. Gofer, and D. Aurbach, *J. Electrochem. Soc.* 151, A1044-A1051 (2004).
7. Comparative study of  $Mg^{2+}$  and  $Li^+$  ion insertions into the  $Mo_6S_8$  Chevrel phase using electrochemical impedance spectroscopy, M. D. Levi, **H. Gizbar**, E. Lancry, Y. Gofer, E. Levi, and D. Aurbach, *J. Electroanal. Chem.* 569, 211-223 (2004).
8. Phase transitions and diffusion kinetics during  $Mg^{2+}$  and  $Li^+$  ion insertions into the  $Mo_6S_8$  Chevrel phase compound studied by PITT, M. D. Levi, E. Lancry, **H. Gizbar**, Y. Gofer, E. Levi, and D. Aurbach, *Electrochim. Acta*, 49, 3201-3209 (2004).
9. Alkyl group transmetallation reactions in electrolytic solutions studied by multinuclear NMR, **H. Gizbar**, Y. Viestfrid, O. Chusid, Y. Gofer, H. E. Gottlieb, V. Marks, and D. Aurbach, *Organometallics* 23, 3826-3831 (2004).

10. The effect of the anionic framework of  $\text{Mo}_6\text{X}_8$  Chevrel Phase ( $\text{X}=\text{S}, \text{Se}$ ) on the thermodynamics and the kinetics of the electrochemical insertion of  $\text{Mg}^{+2}$  ions, Levi-MD, Lancri-E, Levi-E, **Gizbar-H**, Gofer-Y, Aurbach-D. *Solid State Ionics Diffusion and-Reactions*, 176(19-22), 1695-9 June (2005).
11. Improved electrolyte solutions for rechargeable magnesium batteries, Gofer-Y, Chusid-O, **Gizbar-H**, Viestfrid-Y, Gottlieb-HE, Marks-V, Aurbach-D. *Electrochemical-and-Solid-State-Letters*, 9(5), A257-60 (2006).
12. Unique Behavior Of Dimethoxyethane (DME) /  $\text{Mg}(\text{N}(\text{SO}_2\text{CF}_3)_2)_2$  Solutions, M. Salama, I. Shterenberg, **H. Gizbar**, N. Eliad, M. Kosa, K. Adamsky, M. Afri, L. shimon, H. Gottlib, D. Major, Y. Gofer, D. Aurbach, *J. Physical Chemistry C* 120(35), 19586-19594, 2016.
13. Offer Zeiri, Noa Fruchter, Eyal Elish, **Haim Gizbar**, Dror Shamir, Itzhak Sedgi, "Determination of uranium isotopic ratio by ICP-OES using optimal sensitivity position analysis", *Analytical Chemistry*, 93, 5123, (2021).

Patents

14. High energy rechargeable electrochemical cells with nonaqueous electrolytes (rechargeable magnesium batteries). D. Aurbach, L. Zhongua, A. Schechter, Y. Gofer and **H. Gizbar**. U.S. patent No. 6,316,141, Nov. 2001.
15. High energy, rechargeable batteries with nonaqueous gel polymer electrolyte, an intercalation cathode and magnesium anode. **US Pat. Appl. (2001). Cont. in part of US 6316141.**